Menu	
<ul> <li>Home Home</li> <li>About this portal</li> <li>Latest updates</li> </ul>	
Print Save Email Resource detail Citations  BS 6068-2.36:1989 Water quality - Physical, chemical and biochemical method	de
- Method 2.36: Spectrometric method for the determination of nitrate using sulphosalicylic acid	us
View on Information Provider website {{ linkText }}	
Abbreviation BS 6068-2.36:1989 Valid from 29/09/1989	
Information provider Standards New Zealand Author British Standards Institution Information type British Standard Format PDF	
Cited By  This resource is cited by 2 documents (show Citations)	
Description	
Applicable to raw and potable water samples containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of the containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using nitrogen from about 0.01 mg/l to 0.2 mg/l using nitrogen from about 0.01 mg/l to 0.2 mg/l using nitrogen from about 0.01 mg/l using nitrogen from 0.01 mg/l usi	of
For assistance with locating previous versions, please contact the information provider.	
View on Information Provider website {{ linkText }}	
For assistance with locating previous versions, please contact the information provider.	

BS 6068-2.36:1989 Water quality - Physical, chemical and biochemical methods - Method 2.36: Spectrometric method for the determination of nitrate using sulphosalicylic acid

This resource is cited by:

Skip to main content Skip to primary navigation

• AS/NZS 4020:2005

BS 6068-2.36:1989 is cited by AS/NZS 4020:2005 Testing of products for use in contact with drinking water

• BS 6920-2.6:2000

BS 6068-2.36:1989 is cited by BS 6920-2.6:2000 Suitability of non-metallic products for use in contact with water intended for human consumption with regard to their effect on the quality of water. Methods of test. The extraction of metals



## BS 6068-2.36:1989 Water quality - Physical, chemical and biochemical methods - Method 2.36: Spectrometric method for the determination of nitrate using sulphosalicylic acid

Show what documents this resource is CITED BY Show what documents this resource CITES

Description

Applicable to raw and potable water samples containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of 25 ml.

View on Information Provider website

BS 6068-2.36:1989 Water quality - Physical, chemical and biochemical methods - Method 2.36: Spectrometric method for the determination of nitrate using sulphosalicylic acid

Description

Applicable to raw and potable water samples containing nitrate nitrogen from about 0.01 mg/l to 0.2 mg/l using a test portion of 25 ml.

View on Information Provider website

This resource does not cite any other resources.

## BS 6068-2.36:1989 Water quality - Physical, chemical and biochemical methods - Method 2.36: Spectrometric method for the determination of nitrate using sulphosalicylic acid

This resource does not CITE any other resources.	
--	--



## **Table of Contents**

Print Save Email		
<u>Feedback</u>		
		_

- Contact us
- Privacy policy
- Disclaimer
- Copyright

	_	·	

<u>Feedback</u>