

Menu

- [Home](#)
- [About this portal](#)
- [Latest updates](#)

Print

[Save](#)

Email

[Resource detail](#)

[Citations](#)

Standard Methods for the Examination of Water and Wastewater - Part 4500-H⁺: PH Value (2012)

[View on Information Provider website](#)

Abbreviation

4500-H⁺: PH Value (2012)

Version

22nd Edition - 2012

Valid from

05/12/2012

Information provider

IHS Markit

Author

American Public Health Association, American Waterworks Association and Water Environment Federation

Information type

Other Standard

Format

PDF

Cited By

[This resource is cited by 1 document \(show Citations\)](#)

Description

Part 4500-H⁺ pH Value

Measurement of pH is one of the most important and frequently used tests in water chemistry.

Following an introductory discussion (A), one method is given for the measurement of pH. This electrometric method (B) is described in detail. A general discussion of the principle on which the method is based, and interferences encountered, is followed by a description of apparatus and procedure, including sections on troubleshooting and precision and bias. A table gives the weight of chemicals needed per 1000 mL aqueous solution at 25°C to prepare various pH standard solutions. A second table gives standard pH values at various temperatures, including primary and secondary standards.

For assistance with locating previous versions, please contact the information provider.

[View on Information Provider website](#)

For assistance with locating previous versions, please contact the information provider.

This resource is cited by:

Standard Methods for the Examination of Water and Wastewater - Part 4500-H+: PH Value (2012)

This document is CITED BY:

- [AS/NZS 2280: 2014](#)

4500-H+: PH Value (2012) is cited by AS/NZS 2280: 2014 Ductile iron pipes and fittings

Standard Methods for the Examination of Water and Wastewater - Part 4500-H+: PH Value (2012)

Description

Part 4500-H⁺ pH Value

Measurement of pH is one of the most important and frequently used tests in water chemistry. Following an introductory discussion (A), one method is given for the measurement of pH. This electrometric method (B) is described in detail. A general discussion of the principle on which the method is based, and interferences encountered, is followed by a description of apparatus and procedure, including sections on troubleshooting and precision and bias. A table gives the weight of chemicals needed per 1000 mL aqueous solution at 25°C to prepare various pH standard solutions. A second table gives standard pH values at various temperatures, including primary and secondary

standards.

[View on Information Provider website](#)

[Standard Methods for the Examination of Water and Wastewater - Part 4500-H+: PH Value \(2012\)](#)

Description

Part 4500-H⁺ pH Value

Measurement of pH is one of the most important and frequently used tests in water chemistry. Following an introductory discussion (A), one method is given for the measurement of pH. This electrometric method (B) is described in detail. A general discussion of the principle on which the method is based, and interferences encountered, is followed by a description of apparatus and procedure, including sections on troubleshooting and precision and bias. A table gives the weight of chemicals needed per 1000 mL aqueous solution at 25°C to prepare various pH standard solutions. A second table gives standard pH values at various temperatures, including primary and secondary standards.

[View on Information Provider website](#)

This resource does not cite any other resources.

Standard Methods for the Examination of Water and Wastewater - Part 4500-H+: PH Value (2012)

This resource does not CITE any other resources.

Table of Contents

[Save](#)

[Feedback](#)

- [Contact us](#)
- [Privacy policy](#)
- [Disclaimer](#)
- [Copyright](#)

--	--	--

[Feedback](#)